R.R.B.

where B = (2, 4) and C = (6, 10), then what is 'k'?

2) 3

1.

1)6

JE Question paper 2010

Based on Memory

If the point A (7, k) is the vertex of an isosceles triangle ABC with base BC,

3) 4

4) 5

2.	If the distance between the points (na, nb) and (a, b) is 4 times the distance				
	between the points (5a, 5b) and (a, b), then 'n' is equal to-				
	a) 11 or -13	2) 11	3) 13	4) 17 or -15	
3.	ABC is a tringle wh	nose centroid is G. I	f A is (-3, 1) B is (2	, b), C is (a, -4) and	
	G is $(1, -1)$ then find	nd 'a' and 'b'.			
	1) $a = 4$, $b = 0$		2) $a = 0$, $b = 4$		
	3) $a = 3$, $b = 2$		4) $a = 5$, $b = 2$		
4.	An angle is equal to	$\frac{3\pi}{5}$ radians. What	is its measure in deg	grees?	
	1) 145°	2) 72°	3) 108°	4) 120°	
5.	The equation of a s	traight line is 2x-3y	+2 = 0. What is its sl	lope?	
	1) $\frac{2}{3}$	2) –2	3) 2	$4) - \frac{2}{3}$	
6.	Find the range of va	alues of x, which sa	tisfy the inequality-		
	$-\frac{1}{5} \le \frac{3x}{10} + 1 <$	$\frac{2}{5}$, $x \in R$			
	1) $(x : x \in R, 0.3 \le$	x < 9)	2) $(x : x \in R, -4 \le$	(x < -2)	
	3) $(x : x \in R, 4 \ge x)$	> -2)	4) $(x : x \in R, 5 < x)$	$x \le 8$)	

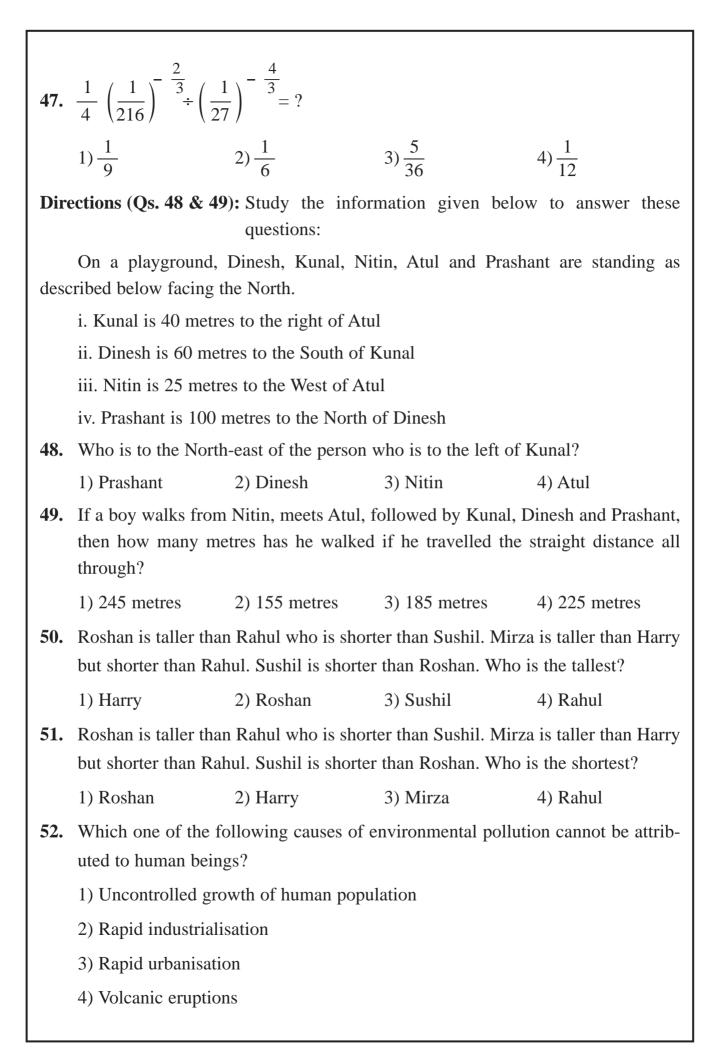
7.	Read the law given below and identify the same:				
	The mass on any substance liberated from an electrolyte is directly proportional				
	to the quantity of charge passing through the solution.				
	1) Avogadro's law				
	2) Faraday's first la	w of electrolysis			
	3) Faraday's second	l law of electrolysis			
	4) Kirchhaoff's law	of electricity			
8.	The value of Avoga	adro's constant is-			
	1) 6.022×10^{23} per	mole	2) 58.04×10^{-2} pc	er mole	
	3) 69.51×10^{-18} p	er mole	4) 6.022×10^{14} pc	er mole	
9.	In an experiment,	295 mg of copper	is deposited when a	a current of 500 mA	
	_		chemical equivalent		
	1) 32.77 a 10 ⁻⁸ kg	/ coulomb	2) 58.4 kg/ coulon		
	3) $109.5 \times 10^8 \text{ kg/}$	coulomb	$4. \frac{1}{32.77 \times 10^{-8}}$	kg/ coulomb	
10.	Which one of the fe	ollowing is the corre	ect unit of angular ve	elocity?	
	1) m/ minute	$2) \text{ cm/ } \text{sec}^2$	3) cm/sec	4) radians/ sec	
11.	The force by which a body is attracted towards the centre of the earth is called-				
	1) Gravitational for	ce	2) Mass		
	3) Momentum		4) Impulsive force	;	
12.	The maximum disp	lacement of a vibrat	ing body from its me	ean position is called-	
	1) Gyration	2) Wavelength	3) Amplitude	4) Impulse	
13.	The kinetic energy	of a body depends to	upon-		
	1) Mass, gravity an	d height	2) Its mass alone		
	3) Its velocity alone	e	4) Both mass and	velocity	
14.	A ball weighing 25 grams is thrown vertically into the air. It takes 15 seconds to reach its highest point. How much time would it take to reach the ground from its highest point?				
	1) More data are re	quired for calculation	on		
	2) Less than 15 sec	onds			
	3) More than 15 se	conds			
	4) 15 seconds				
I					

15.	The term 'Squirrel Cage' is associated with			
	1) Pressure gauges		2) Internal combus	tion engines
	3) Potentiometers		4) Electric motors	
16.	The phenomenon of	f increase in the ten	perature of the earth	's atmosphare due to
	absorption of the ir	nfra-red radiations re	eflected from the eart	th's surface is called-
	1) Tsunami		2) Solar heating	
	3) Green-house effe	ect	4) Seismic effect	
17.	Why is it recomm closed rooms?	ended that people s	hould not use charc	oal or gas stoves in
	1) The electrical wa	iring in the room ma	ny catch fire	
	2) The stoves will a	get extinguished		
	3) It can cause carb	oon monoxide poiso	ning	
	4) The stoves may	burst		
18.	The most effective	way to improve saf	ety in a vast organis	ation like the Indian
	Railways is to			
	1) Ignore small act	s of negligence by the	ne staff	
	2) Carry out frequa	ant checks		
	3) Educate the staff	f at all levels		
	4) Punish defaultin	g staff		
19.	The density of water	er is maximum at		
	1) 100°C	2) 0°C	3) -273°C	4) 4°C
20.	Which one of the f	ollowing quantities	does not have a unit?	?
	1) Velocity	2) Density	3) Specific Gravity	4) Mass
21.	A Swimmer finds	it easier to swim in	sea water than in pla	in water. Why?
	1) Sea water has le	ss contamination		
	2) Sea waves help	a swimmer to swim		
	3) Sea water has hi	gher density than pl	ain water	
	4) Sea has a much	higher volume of wa	ater	

22.	Humidity refers to-				
	1) Both temperature and moisture contents of the air				
	2) Temperature of the air				
	3) Moisture content	t of the air			
	4) Presure of the air	r			
23.	Boyle's law states t	hat-			
	1) Volume is direct	ly proportional to te	mperature		
	2) Pressure is inver	sely proportional to	temperature		
	3) Pressure is direct	tly proportional to te	emperature		
	4) Presure is invers	ely proportional to v	valume		
24.	Purity of milk is co	onfirmed by-			
	1) Barometer	2) Lactometer	3) Altimeter	4) Hygroscope	
25.	A stick is dipped in a vessel containing water. It appears bent due to the property of-				
	1) Reflection 2) Newton's Law of Motion				
	3) Refraction		4) Buoyancy		
26.	The temperature on	the surface of the s	un is about-		
	1) 8×10^{15} °C	2) 500°C	3) 6000°C	4) 1000°C	
27.	The planet farthest	from the Sun is-			
	1) Pluto	2) Mercury	3) Jupiter	4) Neptune	
28.	Which one of the fo	ollowing is measured	d on the 'RICHTER	SCALE'?	
	1) The speed of a re	ocket 5 seconds after	r take off		
	2) The intensity of	thunderstorm			
	3) The intensity of	an earthquake			
	4) The speed at wh	ich a player serves t	he ball in Lawn Teni	nis	
29.			or shrillness of its w	histle increases. This	
	phenomenon is exp	-			
	1) Big Bang Theory	y	2) Doppler Effect		
	3) Charles' Law		4) Archimedes Prin	nciple	

30.	The load on a spring per unit deflection is called-				
	1) Stress	2) Flexbility	3) Stiffness	4) Strain	
31.	The term acceleration	on means-			
	1) Maximum speed	l of a vehicle	2) Rate of change	of time	
	3) Rate of change of	of velocity	4) Rate of change	of distance	
32.	A body of mass 10 would the body tra		rest at the rate of 3 n	n/sec ² . What distance	
	1) 250 metres	2) 100 metres	3) 150 metres	4) 200 metres	
33.	•	•	0%. If 10,000 joules are by the engine wou	s of heat energy are	
	1) 40,000 Joules	2) 10,000 Joules	3) 25,000 Joules	4) 4,000 Joules	
34.	A gas is allowed to expand at constant temperature from an initial volume of 10 ml to a final volume of 300 ml. At the end of the expansion, the pressure of the gas was found to be 1 atmosphere. What was the initial pressure of the gas?				
	1) 9 atmosphere	2) 1 atmosphere	3) 3 atmosphere	4) $\frac{1}{3}$ atmosphere	
35.	There are three no through them?	n-collinear points. I	How many circles co	an be drawn passing	
	1) Infinite	2) One	3) Two	4) Three	
36.	What do you under	estand by the term 'A	Absolute Pressure'?		
	1) It is the atmosph	neric pressure at mea	an sea level		
	2) It is the atmosph	neric pressure expres	ssed in kg/ cm ²		
	3) It is the pressure sures	equal to the algebra	aic sum of atmosphe	ric and gauge pres	
	4) It is the pressure	as seen on the gaug	ge of a pressure measure	suring instrument	
Dire	ections (Qs. 37 to 39	9): Study the folloi	wng number seque	nce to answer these	
		questions.			
		63158638522			
37.	How many odd nur odd number?	mbers in the above s	sequence are immedi	ately followed by an	
1					

38.	How many even numbers are there in the sequence which are immediately preceded by an odd number but immediately followed by an even number?				
	1) 5	2) 2	3) 3	4) 4	
39.	How many odd no		n the sequence whi d by an even number	ch are immediately	
	1) 5	2) 2	3) 3	4) 4	
40.	Study the following	number sequence-			
	5 9 8 1 3 2 7 4 3 8				
		e fifth and sixth dig	-	anged, also the third which digit would be	
	1) 8	2) 1	3) 4	4) 7	
41.	If the numbers from 1 to 45 which are exactly divisible by 3 are arranged in an ascending order, minimum number being kept frist, then which number would come at the ninth place from the first?				
	1) 30	2) 21	3) 24	4) 27	
42.	Find the value of-				
	$8.55 \times 8.55 - 2 \times 8$	$3.55 \times 3.55 + 3.55 \times$	3.55		
	1) 27.5	2) 20	3) 25	4) 36	
43.		e have six married s f members in the far		m has four children.	
	1) 40	2) 30	3) 36	4) 38	
Dire	ections (Qs. 44 to 46	6): In each of the let	ter series given in th	nese questions, some	
			9	order as one of the	
	rntives below it. Cho	ose the correct alter	native.		
44.	ba-b-aab-a-b				
	1) babb	2) abab	3) abba	4) baba	
45.	mnonopqopqrs				
	1) qrstu	2) mnopq	3) oqrst	4) pqrst	
46.	c-bba-cab-ac-ab	-ac			
	1) bcacb	2) abcbc	3) acbcb	4) babcc	



53.	Which one of the following gases is manly responsible for the GREENHOUSE EFFECT?			
	1) Sulphur dioxide		2) Carbon mono-ox	xide
	3) Hydrogen sulphi	ide	4) Carbon dioxide	
54.	Which one of the f	ollowing is a major	constituent of petrol:	?
	1) Pentane (C_5H_{12}))	2) Octane (C ₈ H ₁₈)	
	3) Methane (CH ₄)		4) Hexane (C ₆ H ₁₄))
55.	Which one of the f	ollowing is a widely	used solid lubricant	?
	1) Graphite	2) Sodium	3) Lithium	4) Zinc
56.	The world TSUNA	MI is derived from	which of the following	ng languages?
	1) Sinhalese	2) Korean	3) Chinese	4) Japanese
57.	'. A major nuclear power plant, located in one of the countries affected TSUNAMI, escaped damage. Where is it located?			ountries affected by
	1) Bali in Indonesia	a	2) Galle in Sri Lanka	
	3) Phuket in Thaila	nd	4) Kalpakkam in Ir	ndia
58.	A major cricket g Where is it locted?	round was severely	damanged by the	rescent TSUNAMI.
	1) Candy in Sri Lar	nka	2) Chittagong in Ba	angladesh
	3) Galle in Sri Lan	ka	4) Nairobi in Keny	a
59.	The sound waves in	n the audible range l	nave frequencies in the	he range of-
	1) 20 Hz to 20,000	Hz	2) 0.5 Hz to 5 Hz	
	3) 1 Hz to 10 Hz		4) 20,000 hz to 40,	000 Hz
60.	Which of the following being used for application such as assessing depth oceans, thickness measurement, determination of the position of icebergs, fla detection in metals, etc?			
	1) Ultrasonic wave	s 2) X–rays	3) Light waves	4) γ–rays
61.	The isotopes of an	element are characte	erised by which of th	e following?
	1) Presence of neut	rons of unusual size		
	2) Different number	er of electrons in the	atom	
	3) Different number	or of protons in the n	ucleus	
	4) Different number of neutrons in the nucleus			

62.	. How do you understand by the term 'Binding	g Energy'?				
0_1	1) Energy released when a nucleus is formed from protons and neutrons					
	2) The force of attraction between an electron in the first orbit and the nucleus					
	3) Electron belonging to the same major energy level					
	4) Energy associated with a photon					
63.		9				
05.	1) Ionic bonds are non-rigid and non-direction					
	2) Compounds formed by ionic bonds are no		f alactricity			
	3) Ionic bonds are formed by transfer of elec-		•			
	atom	cuons nom a m	letai to a non–metai			
	4) Compounds fromed by ionic bonds are ha	ard and brittle				
64.	• Arrange the following materials in the order	of decreasing c	onductivity:			
	Silicon, Glass, Aluminium, Silver					
	1) Glass, Silicon, Aluminium, Silver	1) Glass, Silicon, Aluminium, Silver				
	2) Aluminium, Silver, Glass, Silicon					
	3) Silver, Silicon, Aluminium, Glass	3) Silver, Silicon, Aluminium, Glass				
	4) Silver, Aluminium, Silicon, Glass					
65.	If a barometer carries water instead of mercui	ry, then the heig	tht of the column for			
	a pressure equivalent to 75 cm of mercury w	ould be-				
	1) 1050 cm 2) 1020 cm 3) 1	1000 cm	4) 5.5 cm			
66.	• The term EURO-II in the context of modern	n cars refers to-				
	1) Emission from cars 2) S	Speed of cars				
	3) Fuel efficiency 4) T	Torque available				
67.	. What is the ultimate benefit of good commu	What is the ultimate benefit of good communication in a vast organisation like				
	the Indian Railways?					
	1) Improved productivity and profits					
	2) Reduced frustration among the employees	S				
	3) Development of good human relations					
	4) Improved image of the organisation					

68.	What is the term AGMARK used for?				
	1) Grading various agricultural commodities				
	2) Grading battery t	coys			
	3) Grading polyeste	er textiles			
	4) Grading engine l	ubricating oils			
69.	The standard used i	n India for certifying	g the quality of Indu	strial goods is-	
	1) ISI	2) ISO	3) ITI	4) CEERI	
70.	An electric heater of days, it will consum	•	d to heat water everd	ay for 2 hours. In 10	
	1) 20 kWh	2) 2 kWh	3) 0.2 kWh	4) 200 kWh	
71.	Ozone is a gas havi	ng atoms of Ox	gen in its molecules.		
	1) Four	2) One	3) Two	4) Three	
72.	· ·	s 14.5 Kg Of LPG in age energy consume	•	ific value of LPG is	
	1) 275 kj	2) 27.5 kj	3) 27,500 kj	4) 0.275 kj	
73.	The chemical formu	ıla of natural gas is-			
	1) C_3H_8	2) CH ₄	3) C_4H_{10}	4) C_2H_6	
74.	The percentage of c	arbon in one molecu	ale of carbon dioxide	e is approximately-	
	1) 2.73%	2) 72.7%	3) 80%	4) 27.3%	
75.	The term 'Cracking' in the context of organic molecules is-				
	1) The process of fr	ractional distillation	in the refineries		
	2) Breaking of a large alkane molecule into smaller hydrocarbon molecules				
	3) A nuclear reactio	n where in the nucle	eus is broken		
	4) Use of fire crack	ers to produce heat	to initiate certain che	emical reactions	
76.	In a nuclear power station, which one of the following is commonly used as a fuel for producing heat?				
	1) Coal	2) Helium	3) Heavy Water	4) Uranium-235	
77.				e number of fissions	
	-		f 10 MW for 10 hou		
	1) 6.5×10^{50}	2) 2.1×10^{12}	3) 1.125×10^{22}	4) 1800	
78.				ne calorific value of	
		_	of consumption of the		
	1) 0.1 2) 1.5 3) 1 4) 0.5				

70	TC 1	4	2 4 4	C.1	
19.	If acceleration due to gravity is 10 m/ sec ² , then the potential energy of a body of mass 1 kg kept at a height of 5 metres is-				
	1) 50 Joules	_		4) 10 Joules	
80.	,		,	displaced would be-	
	1) 220 kg	2) 0 kg	3) 180 kg	-	
81.	,	,	,	u cm is dipped in a	
	beaker containing would be reduced by		vity 1 gm/ cu cm. T	he weight of the ball	
	1) Colleting more of	data for making the	calculation		
	2) 0.1 gm				
	3) 1 gm				
	4) 10 gm				
82.	Archimedes Princip	ole is related to-			
	1) laws of floatatio	n	2) Right-angled tr	iangle	
	3) Laws of gravity		4) Relation between current and vo		
83.	• The commonly used washing soda is-				
	1) Sodium Bicarbo	nate	2) Sodium Carbona	ate	
	3) Sodium Chloride	e	4) Magnesium Chl	oride	
84.	The chemical form	ula of 'plaster of par	is' is-		
	1) 2CaSO_4 . $\frac{1}{2}$ H ₂	O	2) Ca(OH) ₂		
	3) (CaSO ₄) ₂ .H ₂ O		4) CaOCl ₂		
85.	A sanitary worker	uses a white substan	nced to clean water	tanks. The substance	
	_	of chlorine. The sub			
	1) Bleaching powd		2) Slaked lime		
	3) Backing powder		4) Common salt		
86.	_			mall in size. Which the cake to rise and	
	1) Cooking oil		2) Baking powder		
	3) Bleaching powd	er	4) Sugar		

87. A White chemical compound becomes hard on mixing proper quantity of water. It is also used in surgery to repair fractured bones. What is it? 1) Plaster of paris 2) Slaked lime 3) Bleaching power 4) lime **88.** Brass has which of the following compositions? 1) 40% copper, 40% zinc and 20% tin 2) 50% zinc and 50% copper 3) 80% zinc, 10% copper and 10% lead 4) 80% copper and 20% zinc **89.** Broneze has which of the following compositions? 1) 50% copper, 10% iron and 40% zinc 2) 90% copper and 10% tin 3) 10% copper and 90% tin 4) 40% copper, 40% tin and 20% zinc **90.** Solder has which of the following compositions? 1) 50% lead and 50% tin 2) 70% lead, 20% copper and 10% tin 3) 20% lead, 40% copper and 40% tin 4) 10% lead and 90% tin **91.** Galvansation is the process of-1) Drawing metals into thin wires 2) Giving a coating of zinc metal on iron 3) Making aluminium metal into thin wire 4) Making thin aluminium foils **92.** German silver has which of the following compositions? 1) 20% copper, 20% chromium and 60% zinc 2) 40% copper, 20% zinc and 40% silver 3) 60% copper, 20% zinc and 20% nickel 4) 80% copper, 10% zinc and 10% silver The symbol of Magnesium is Mg. What does Mg^{2+} mean? 93. 1) Magnesium atom has acquired two protons 2) two atoms of magnesium have combined 3) Magnesium atom has donated two outermost electrons to form a positive ion 4) The charged Mg. ion attracts oppositely charged negative ions with twice as much intensity

94. When Sodium (Na), Copper (Cu) and Zinc (Zn) are placed in the order of decreasing reactivity, then their order would be-1) Na > ZN > Cu2) Na > Cu > Zn 3) Cu > Na > Zn4) Zn > Na > Cu**95.** Which of the following metals is more reactive than Hydrogen? 2) Calcium 3) Aluminium 1) Gold 4) Iron **96.** Which of the following metals can displace Hydrogen from its compounds like water and acids to form hydrogen gas? 1) Tin 2) Copper 3) Mercury 4) Silver 97. The approximate percentage of salt by weight in sea water is-1) 41% 2) 3.6% 3) 0.1% 4) 10.2% 98. The common salt is iodised to prevent occurrence of which of the following diseases in the human body? 1) Diabetes 2) Goitre 3) Beri-beri 4) Night-blindness 99. A wire of a certain length has a resistance of 2.2Ω . If the wire is stretched to twice its original length, then find the new resistence. 2) 1.1Ω $1)~8.8\Omega$ 3) 2.2Ω 4) 4.4Ω 2Ω 3Ω 100. In the above circuit, the effective resitance between the points A and B is- 4Ω 6Ω 2) $4\frac{4}{9} \Omega$ 3) $6\frac{1}{3}\Omega$ 4) $3\frac{1}{3}\Omega$ 1) 18Ω **ANSWERS** 1-4; 2-4; 3-1; 4-3; 5-1; 6-2; 7-2; 8-1; 9-1; 10-4; 11-1; 12-3; 13-4; 14-4; 15-4; 16-3; 17-3 18-2; 19-4; 20-3; 21-3; 22-3; 23-4; 24-2; 25-3; 26-3; 27-1; 28-3; 29-2; 30-3; 31-3; 32-3; 33-1; 34-3; 35-2; 36-4; 37-1; 38-3; 39-3; 40-1; 41-1; 42-3; 43-4; 44-3; 45-4; 46-3; 47-1; 48-1; 49-4; 50-2; 51-2; 52-4; 53-4; 54-2; 55-1; 56-4; 57-4; 58-3; 59-1; 60-1; 61-4; 62-1; 63-1; 64-1; 65-2; 66-1; 67-1; 68-1; 69-1; 70-1; 71-4; 72-3; 73-2; 74-4; 75-2; 76-4; 77-3; 78-3; 79-1; 80-4; 81-4; 82-1; 83-2; 84-3; 85-4; 86-2; 87-1; 88-4; 89-2; 90-1; 91-2; 92-3; 93-3; 94-1; 95-2; 96-1; 97-2; 98-2; 99-1; 100-3.